

[illegible]

1. **Molar mass**
The mass of one mole of a substance, expressed in grams per mole (g/mol). It is calculated by summing the atomic masses of all atoms in the chemical formula.

2. **Significant figures**
The digits in a measurement that are known with certainty, plus one digit that is estimated. They indicate the precision of the measurement.

3. **Solubility**
The ability of a substance to dissolve in a solvent. It is often expressed as the maximum amount of solute that can dissolve in a given amount of solvent at a specific temperature.

4. **Chemical equation**
A symbolic representation of a chemical reaction, showing the reactants on the left and the products on the right, separated by an arrow.

5. **Chemical reaction**
A process in which one or more substances (reactants) are converted into one or more new substances (products) through the breaking and forming of chemical bonds.

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«АЛТАЙСКИЙ ГОСУДАРСТВЕННЫЙ КОЛЛЕДЖ»

ТЕМА: СОВЕРШЕНСТВОВАНИЕ РАБОТЫ ПО
УПРАВЛЕНИЮ ПЕРСОНАЛОМ В ООО "КРАТОН"

Выполнил: _____
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